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(Phenyl)(1-phenylsulfonyl-1*H*-indol-3-yl)methanone

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Key indicators: single-crystal X-ray study; T = 295 K; mean σ (C–C) = 0.003 Å; R factor = 0.037; wR factor = 0.109; data-to-parameter ratio = 18.2.

In the title compound, C₂₁H₁₅NO₃S, the sulfonyl-bound phenyl ring forms a dihedral angle of $86.28(5)^{\circ}$ with the indole ring system. The molecular structure is stabilized by intramolecular $C-H\cdots O$ hydrogen bonds. The crystal packing is stabilized by weak intermolecular C-H···O and $C-H\cdots\pi$ interactions.

Related literature

For the structures of closely related compounds, see: Chakkaravarthi et al. (2007, 2008).



Experimental

Crystal data $C_{21}H_{15}NO_3S$

 $M_r = 361.40$

Triclinic, P1	
a = 7.567 (1) Å	
b = 10.571 (2) Å	
c = 12.083 (3) Å	
$\alpha = 66.302 \ (2)^{\circ}$	
$\beta = 80.740 \ (1)^{\circ}$	
$\gamma = 78.403 \ (1)^{\circ}$	

Data collection

Bruker Kappa APEXII	15638 measured reflections
diffractometer	4276 independent reflections
Absorption correction: multi-scan	3187 reflections with $I > 2\sigma(I)$
(SADABS; Sheldrick, 1996)	$R_{\rm int} = 0.026$
$T_{\rm min} = 0.952, \ T_{\rm max} = 0.960$	Standard reflections: 0

Refinement

$R[F^2 > 2\sigma(F^2)] = 0.037$	2 restraints
$wR(F^2) = 0.109$	H-atom parameters constrained
S = 1.01	$\Delta \rho_{\rm max} = 0.21 \ {\rm e} \ {\rm \AA}^{-3}$
4276 reflections	$\Delta \rho_{\rm min} = -0.29 \text{ e} \text{ Å}^{-3}$
235 parameters	

Table 1 Hydrogen-bond geometry (Å, °).

Cg1 is the centroid of the C16-C21 ring.

$D - H \cdots A$	D-H	$H \cdot \cdot \cdot A$	$D \cdots A$	$D - \mathbf{H} \cdots A$
$C13-H13\cdots O1$ $C6-H6\cdots O2^{i}$ $C7-H7\cdots O2^{i}$ $C4-H4\cdots Cg1^{ii}$	0.93	2.42	2.999 (2)	120
	0.93	2.58	3.493 (2)	167
	0.93	2.54	3.429 (2)	160
	0.93	2.98	3.774 (3)	144

Symmetry codes: (i) -x + 2, -y, -z; (ii) -x + 1, -y, -z.

Data collection: APEX2 (Bruker, 2004); cell refinement: SAINT (Bruker, 2004); data reduction: SAINT; program(s) used to solve structure: SHELXS97 (Sheldrick, 2008); program(s) used to refine structure: SHELXL97 (Sheldrick, 2008); molecular graphics: PLATON (Spek, 2009); software used to prepare material for publication: SHELXL97.

The authors wish to acknowledge DV University of Madras for the data collection.

Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: BT5379).

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organic compounds

V = 863.5 (3) Å³

Mo $K\alpha$ radiation

 $0.24 \times 0.22 \times 0.20 \text{ mm}$

 $\mu = 0.21 \text{ mm}^{-1}$ T = 295 K

7 - 2

supplementary materials

Acta Cryst. (2010). E66, o2895 [doi:10.1107/81600536810042261]

(Phenyl)(1-phenylsulfonyl-1*H*-indol-3-yl)methanone

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Comment

The geometric parameters of the molecule of (I) (Fig. 1) agree well with the reported values of similar structures (Chakkara-varthi *et al.*, 2007; Chakkaravarthi *et al.*, 2008). The phenyl rings C1—C6 and C16—C21 form the dihedral angles of 86.28 (5)° and 51.91 (5)°, respectively with the indole ring system. The mean planes of the two phenyl rings are inclined at an angle of 42.16 (6)°.

The sum of the bond angles around N1 [358.53°] indicates that N1 atom is sp^2 hybridized. The molecular structure is stabilized by intra molecular C—H···O hydrogen bonds and the crystal packing is stabilized by weak intermolecular C—H···O and C—H··· π [C4—H4···Cg1 (1 - x, -y, -z) distance of 3.774 (3)Å (Cg1 is the centroid of the ring defined by the atoms C16—C21)] interactions.

Experimental

To a solution of 1-phenylsulfonyl-(1*H*-indol-3-yl)(phenyl)methanol (1 g, 2.75 mmol) in dry 1,2-Dichloroethane (30 ml), manganese dioxide (6 g. 68.96 mmol) was added then stirred at room temperature for 4 h and then refluxed for 3 h. Then the resulting solution was passed through celite pad and washed with DCM (2 x 30 ml). Removal of solvent followed by crystallization from methanol afforded the compound as a colourless crystal.

Refinement

H atoms were positioned geometrically and refined using riding model with C—H = 0.93 Å and $U_{iso}(H) = 1.2Ueq(C)$ for aromatic C—H. The components of the anisotropic displacement parameters in direction of the bond of C3and C4; C18 and C19 were restrained to be equal within an effective standard deviation of 0.001 using the DELU command in *SHELXL* (Sheldrick, 2008).

Figures



Fig. 1. The molecular structure of (I), with atom labels and 30% probability displacement ellipsoids.



Fig. 2. The packing of (I), viewed down the *a* axis. H-bonds are shown as dashed lines; H atoms not involved in hydrogen bonding have been omitted.

(Phenyl)(1-phenylsulfonyl-1*H*-indol-3-yl)methanone

Crystal data	
C ₂₁ H ₁₅ NO ₃ S	Z = 2
$M_r = 361.40$	F(000) = 376
Triclinic, $P\overline{1}$	$D_{\rm x} = 1.390 {\rm Mg m}^{-3}$
Hall symbol: -P 1	Mo <i>K</i> α radiation, $\lambda = 0.71073$ Å
a = 7.567 (1) Å	Cell parameters from 6196 reflections
b = 10.571 (2) Å	$\theta = 2.2 - 27.1^{\circ}$
c = 12.083 (3) Å	$\mu = 0.21 \text{ mm}^{-1}$
$\alpha = 66.302 \ (2)^{\circ}$	T = 295 K
$\beta = 80.740 \ (1)^{\circ}$	Block, colourless
$\gamma = 78.403 \ (1)^{\circ}$	$0.24 \times 0.22 \times 0.20 \text{ mm}$
$V = 863.5 (3) \text{ Å}^3$	

Data collection

Bruker Kappa APEXII diffractometer	4276 independent reflections
Radiation source: fine-focus sealed tube	3187 reflections with $I > 2\sigma(I)$
graphite	$R_{\rm int} = 0.026$
ω and ϕ scans	$\theta_{\text{max}} = 28.3^{\circ}, \ \theta_{\text{min}} = 1.9^{\circ}$
Absorption correction: multi-scan (SADABS; Sheldrick, 1996)	$h = -10 \rightarrow 10$
$T_{\min} = 0.952, T_{\max} = 0.960$	$k = -14 \rightarrow 13$
15638 measured reflections	$l = -16 \rightarrow 16$
15638 measured reflections	$l = -16 \rightarrow 16$

Refinement

Refinement on F^2	Primary atom site location: structure-invariant direct methods
Least-squares matrix: full	Secondary atom site location: difference Fourier map
$R[F^2 > 2\sigma(F^2)] = 0.037$	Hydrogen site location: inferred from neighbouring sites
$wR(F^2) = 0.109$	H-atom parameters constrained
<i>S</i> = 1.01	$w = 1/[\sigma^2(F_o^2) + (0.0543P)^2 + 0.1447P]$ where $P = (F_o^2 + 2F_c^2)/3$
4276 reflections	$(\Delta/\sigma)_{max} < 0.001$
235 parameters	$\Delta \rho_{max} = 0.21 \text{ e } \text{\AA}^{-3}$

supplementary materials

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$\Delta \rho_{\rm min} = -0.29 \text{ e } \text{\AA}^{-3}$

	x	У	Z	$U_{\rm iso}$ */ $U_{\rm eq}$
C1	0.7233 (2)	0.05722 (16)	0.20701 (13)	0.0494 (4)
C2	0.6154 (3)	0.0869 (2)	0.29978 (15)	0.0648 (5)
H2	0.6523	0.1388	0.3360	0.078*
C3	0.4508 (3)	0.0376 (2)	0.33759 (18)	0.0791 (5)
Н3	0.3759	0.0567	0.3997	0.095*
C4	0.3975 (3)	-0.0394 (2)	0.2838 (2)	0.0806 (6)
H4	0.2866	-0.0718	0.3097	0.097*
C5	0.5061 (3)	-0.0687 (2)	0.1928 (2)	0.0757 (5)
Н5	0.4690	-0.1215	0.1575	0.091*
C6	0.6701 (2)	-0.02074 (19)	0.15289 (17)	0.0615 (4)
H6	0.7440	-0.0403	0.0906	0.074*
C14	0.80636 (19)	0.40437 (16)	0.04797 (13)	0.0448 (3)
C13	0.7914 (2)	0.44779 (18)	0.14366 (15)	0.0547 (4)
H13	0.8305	0.3873	0.2188	0.066*
C12	0.7164 (2)	0.5838 (2)	0.12251 (18)	0.0641 (5)
H12	0.7018	0.6157	0.1853	0.077*
C11	0.6617 (3)	0.6751 (2)	0.00966 (19)	0.0681 (5)
H11	0.6130	0.7671	-0.0019	0.082*
C10	0.6778 (2)	0.63262 (18)	-0.08489 (17)	0.0609 (4)
H10	0.6412	0.6948	-0.1603	0.073*
C9	0.75050 (19)	0.49375 (16)	-0.06607 (14)	0.0472 (3)
C8	0.7882 (2)	0.41551 (16)	-0.14359 (13)	0.0474 (3)
C7	0.86292 (19)	0.28383 (16)	-0.07641 (12)	0.0469 (3)
H7	0.8982	0.2113	-0.1040	0.056*
C15	0.7413 (2)	0.46868 (18)	-0.26928 (14)	0.0546 (4)
C16	0.8200 (2)	0.39033 (18)	-0.34881 (13)	0.0553 (4)
C21	0.9990 (3)	0.32957 (19)	-0.35204 (14)	0.0634 (5)
H21	1.0737	0.3291	-0.2978	0.076*
C20	1.0679 (3)	0.2692 (2)	-0.43577 (17)	0.0807 (6)
H20	1.1893	0.2301	-0.4391	0.097*
C19	0.9549 (4)	0.2676 (2)	-0.51451 (18)	0.0891 (6)
H19	1.0000	0.2259	-0.5700	0.107*
C18	0.7771 (4)	0.3274 (3)	-0.51087 (19)	0.0900 (6)
H18	0.7017	0.3266	-0.5642	0.108*
C17	0.7103 (3)	0.3879 (2)	-0.42929 (16)	0.0742 (5)
H17	0.5894	0.4283	-0.4275	0.089*
N1	0.87874 (17)	0.27365 (13)	0.04012 (10)	0.0461 (3)
O1	0.98538 (16)	0.15404 (13)	0.24616 (10)	0.0602 (3)
O2	1.04747 (14)	0.03484 (12)	0.10209 (10)	0.0543 (3)
O3	0.63449 (19)	0.57703 (15)	-0.30880 (11)	0.0789 (4)
S1	0.92933 (5)	0.12195 (4)	0.15505 (3)	0.04628 (13)

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (\hat{A}^2)

Atomic displacement parameters $(Å^2)$

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
C1	0.0554 (8)	0.0428 (8)	0.0381 (7)	0.0021 (7)	-0.0076 (6)	-0.0064 (6)
C2	0.0723 (11)	0.0626 (11)	0.0470 (9)	-0.0010 (9)	0.0007 (8)	-0.0146 (8)
C3	0.0696 (12)	0.0795 (14)	0.0578 (11)	0.0037 (10)	0.0110 (9)	-0.0082 (9)
C4	0.0579 (10)	0.0687 (13)	0.0827 (14)	-0.0089 (9)	-0.0036 (10)	0.0032 (9)
C5	0.0669 (11)	0.0698 (13)	0.0848 (14)	-0.0156 (10)	-0.0104 (10)	-0.0201 (11)
C6	0.0619 (10)	0.0604 (11)	0.0593 (10)	-0.0069 (8)	-0.0060 (8)	-0.0209 (8)
C14	0.0447 (7)	0.0433 (8)	0.0444 (7)	-0.0093 (6)	-0.0023 (6)	-0.0140 (6)
C13	0.0606 (9)	0.0553 (10)	0.0513 (9)	-0.0105 (8)	-0.0034 (7)	-0.0234 (8)
C12	0.0666 (10)	0.0632 (12)	0.0739 (12)	-0.0113 (9)	-0.0011 (9)	-0.0391 (10)
C11	0.0712 (11)	0.0486 (10)	0.0844 (13)	-0.0044 (8)	-0.0049 (10)	-0.0280 (10)
C10	0.0611 (10)	0.0473 (9)	0.0632 (10)	-0.0049 (8)	-0.0074 (8)	-0.0107 (8)
C9	0.0440 (7)	0.0465 (8)	0.0456 (8)	-0.0096 (6)	-0.0016 (6)	-0.0115 (6)
C8	0.0483 (8)	0.0496 (9)	0.0386 (7)	-0.0083 (6)	-0.0041 (6)	-0.0105 (6)
C7	0.0535 (8)	0.0499 (9)	0.0359 (7)	-0.0072 (7)	-0.0052 (6)	-0.0149 (6)
C15	0.0546 (8)	0.0571 (10)	0.0403 (8)	-0.0080 (7)	-0.0082 (7)	-0.0053 (7)
C16	0.0677 (10)	0.0550 (10)	0.0337 (7)	-0.0120 (8)	-0.0094 (7)	-0.0041 (7)
C21	0.0728 (11)	0.0691 (12)	0.0399 (8)	-0.0044 (9)	-0.0087 (8)	-0.0138 (8)
C20	0.1019 (15)	0.0786 (14)	0.0471 (10)	0.0032 (12)	0.0000 (10)	-0.0194 (9)
C19	0.1479 (17)	0.0736 (14)	0.0441 (9)	-0.0172 (13)	-0.0036 (12)	-0.0223 (10)
C18	0.1306 (15)	0.0951 (17)	0.0512 (10)	-0.0334 (13)	-0.0220 (12)	-0.0222 (11)
C17	0.0873 (13)	0.0821 (14)	0.0465 (9)	-0.0182 (11)	-0.0203 (9)	-0.0097 (9)
N1	0.0566 (7)	0.0434 (7)	0.0357 (6)	-0.0037 (6)	-0.0071 (5)	-0.0132 (5)
01	0.0728 (7)	0.0643 (7)	0.0457 (6)	-0.0020 (6)	-0.0197 (5)	-0.0220 (5)
O2	0.0551 (6)	0.0540 (7)	0.0498 (6)	0.0039 (5)	-0.0072 (5)	-0.0205 (5)
03	0.0817 (9)	0.0791 (9)	0.0539 (7)	0.0174 (7)	-0.0184 (6)	-0.0119 (7)
S 1	0.0526 (2)	0.0470 (2)	0.03579 (19)	0.00011 (16)	-0.00991 (15)	-0.01356 (15)

Geometric parameters (Å, °)

C1—C2	1.381 (2)	C10—H10	0.9300
C1—C6	1.386 (2)	С9—С8	1.444 (2)
C1—S1	1.7498 (17)	C8—C7	1.356 (2)
C2—C3	1.384 (3)	C8—C15	1.468 (2)
С2—Н2	0.9300	C7—N1	1.3913 (18)
C3—C4	1.375 (3)	С7—Н7	0.9300
С3—Н3	0.9300	C15—O3	1.227 (2)
C4—C5	1.364 (3)	C15—C16	1.490 (2)
C4—H4	0.9300	C16—C21	1.379 (2)
C5—C6	1.376 (3)	C16—C17	1.387 (2)
С5—Н5	0.9300	C21—C20	1.385 (3)
С6—Н6	0.9300	C21—H21	0.9300
C14—C13	1.387 (2)	C20—C19	1.385 (3)
C14—C9	1.397 (2)	C20—H20	0.9300
C14—N1	1.4152 (19)	C19—C18	1.369 (3)
C13—C12	1.370 (2)	С19—Н19	0.9300

С13—Н13	0.9300	C18—C17	1.361 (3)
C12—C11	1.388 (3)	C18—H18	0.9300
С12—Н12	0.9300	С17—Н17	0.9300
C11—C10	1.366 (3)	N1—S1	1.6677 (12)
C11—H11	0.9300	O1—S1	1.4195 (11)
С10—С9	1.399 (2)	O2—S1	1.4195 (11)
C2—C1—C6	121.26 (17)	С7—С8—С9	107.38 (13)
C2—C1—S1	119.33 (14)	C7—C8—C15	127.21 (15)
C6—C1—S1	119.39 (12)	C9—C8—C15	125.29 (15)
C1—C2—C3	118.46 (19)	C8—C7—N1	109.68 (13)
C1—C2—H2	120.8	С8—С7—Н7	125.2
C3—C2—H2	120.8	N1—C7—H7	125.2
C4—C3—C2	120.39 (18)	O3—C15—C8	119.99 (16)
С4—С3—Н3	119.8	O3—C15—C16	119.61 (14)
С2—С3—Н3	119.8	C8—C15—C16	120.39 (14)
C5—C4—C3	120.5 (2)	C21—C16—C17	118.93 (18)
C5—C4—H4	119.7	C21—C16—C15	123.05 (15)
C3—C4—H4	119.7	C17—C16—C15	117.84 (17)
C4—C5—C6	120.5 (2)	C16—C21—C20	120.24 (18)
С4—С5—Н5	119.8	C16—C21—H21	119.9
С6—С5—Н5	119.8	C20—C21—H21	119.9
C5—C6—C1	118.91 (18)	C21—C20—C19	119.5 (2)
С5—С6—Н6	120.5	C21—C20—H20	120.2
C1—C6—H6	120.5	С19—С20—Н20	120.2
C13—C14—C9	122.50 (15)	C18—C19—C20	120.2 (2)
C13—C14—N1	130.70 (14)	С18—С19—Н19	119.9
C9—C14—N1	106.78 (13)	С20—С19—Н19	119.9
C12—C13—C14	117.03 (16)	C17—C18—C19	120.1 (2)
C12—C13—H13	121.5	C17—C18—H18	120.0
C14—C13—H13	121.5	C19—C18—H18	120.0
C13—C12—C11	121.58 (17)	C18—C17—C16	121.0 (2)
C13—C12—H12	119.2	С18—С17—Н17	119.5
C11—C12—H12	119.2	С16—С17—Н17	119.5
C10-C11-C12	121.37 (17)	C7—N1—C14	108.35 (12)
C10-C11-H11	119.3	C7—N1—S1	123.27 (10)
C12—C11—H11	119.3	C14—N1—S1	126.91 (10)
C11—C10—C9	118.68 (17)	O2—S1—O1	120.91 (7)
C11—C10—H10	120.7	O2—S1—N1	105.43 (7)
C9—C10—H10	120.7	O1—S1—N1	106.66 (7)
C14—C9—C10	118.82 (15)	O2—S1—C1	108.92 (7)
C14—C9—C8	107.78 (14)	O1—S1—C1	109.22 (7)
С10—С9—С8	133.37 (15)	N1—S1—C1	104.41 (7)
C6-C1-C2-C3	-0.4 (3)	C8—C15—C16—C21	-42.0 (2)
S1—C1—C2—C3	178.44 (13)	O3—C15—C16—C17	-35.9 (2)
C1—C2—C3—C4	0.2 (3)	C8—C15—C16—C17	142.89 (16)
C2—C3—C4—C5	0.2 (3)	C17—C16—C21—C20	1.0 (3)
C3—C4—C5—C6	-0.5 (3)	C15—C16—C21—C20	-174.08 (16)
C4—C5—C6—C1	0.3 (3)	C16—C21—C20—C19	-1.4 (3)

supplementary materials

C2—C1—C6—C5	0.2 (3)	C21—C20—C19—C18	1.1 (3)
S1—C1—C6—C5	-178.69 (14)	C20-C19-C18-C17	-0.4 (4)
C9—C14—C13—C12	0.7 (2)	C19—C18—C17—C16	0.0 (3)
N1-C14-C13-C12	178.95 (15)	C21—C16—C17—C18	-0.3 (3)
C14—C13—C12—C11	-1.5 (3)	C15—C16—C17—C18	175.06 (18)
C13-C12-C11-C10	1.0 (3)	C8—C7—N1—C14	1.51 (16)
C12-C11-C10-C9	0.3 (3)	C8—C7—N1—S1	168.55 (10)
C13-C14-C9-C10	0.5 (2)	C13—C14—N1—C7	-179.63 (15)
N1-C14-C9-C10	-178.11 (13)	C9—C14—N1—C7	-1.22 (16)
C13—C14—C9—C8	179.08 (14)	C13—C14—N1—S1	13.9 (2)
N1-C14-C9-C8	0.51 (16)	C9—C14—N1—S1	-167.65 (10)
C11-C10-C9-C14	-1.0 (2)	C7—N1—S1—O2	33.46 (13)
C11—C10—C9—C8	-179.16 (16)	C14—N1—S1—O2	-161.98 (12)
C14—C9—C8—C7	0.40 (16)	C7—N1—S1—O1	163.18 (12)
C10-C9-C8-C7	178.74 (17)	C14—N1—S1—O1	-32.27 (14)
C14—C9—C8—C15	176.58 (14)	C7—N1—S1—C1	-81.25 (13)
C10-C9-C8-C15	-5.1 (3)	C14—N1—S1—C1	83.30 (13)
C9—C8—C7—N1	-1.18 (17)	C2—C1—S1—O2	155.20 (12)
C15—C8—C7—N1	-177.26 (14)	C6—C1—S1—O2	-25.92 (15)
C7—C8—C15—O3	161.36 (17)	C2-C1-S1-O1	21.21 (15)
C9—C8—C15—O3	-14.1 (2)	C6—C1—S1—O1	-159.91 (13)
C7—C8—C15—C16	-17.5 (2)	C2—C1—S1—N1	-92.56 (13)
C9—C8—C15—C16	167.13 (14)	C6—C1—S1—N1	86.32 (14)
O3-C15-C16-C21	139.19 (18)		

Hydrogen-bond geometry (Å, °)

Cg1 is the centroid of the C16–C21 ring.

D—H···A	<i>D</i> —Н	$H \cdots A$	$D \cdots A$	$D\!\!-\!\!\mathrm{H}^{\ldots}\!A$
С2—Н2…О1	0.93	2.59	2.934 (2)	103
С10—Н10…ОЗ	0.93	2.57	3.068 (3)	114
С13—Н13…О1	0.93	2.42	2.999 (2)	120
C6—H6···O2 ⁱ	0.93	2.58	3.493 (2)	167
C7—H7···O2 ⁱ	0.93	2.54	3.429 (2)	160
C4—H4…Cg1 ⁱⁱ	0.93	2.98	3.774 (3)	144
Symmetry codes: (i) $-x+2, -y, -z$; (ii) $-x+1, -y, -z$.				



Fig. 1

Fig. 2

